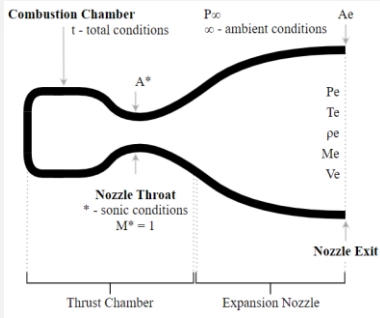


Isentropic Nozzle Flow

Total-to-Static Relations:



$$\frac{T_t}{T} = \left[ 1 + \frac{\gamma-1}{2} M^2 \right]$$

$$\frac{p_t}{p} = \left[ 1 + \frac{\gamma-1}{2} M^2 \right]^{\frac{\gamma}{\gamma-1}}$$

$$\frac{\rho_t}{\rho} = \left[ 1 + \frac{\gamma-1}{2} M^2 \right]^{\frac{1}{\gamma-1}}$$

Mass Flow Rate:

$$\dot{m}(x) = \dot{m}^* = \rho^* u^* A^*$$

$$\dot{m}_p = \frac{p_t A^*}{\sqrt{R T_t}} \left[ \gamma \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}} \right]^{\frac{1}{2}}$$

^This is valid for any choked nozzle.

Isentropic Area-Mach Number Relation:

$$\frac{A}{A^*} = \frac{1}{M} \left\{ \left( \frac{2}{\gamma+1} \right) \left[ 1 + \frac{\gamma-1}{2} M^2 \right] \right\}^{\frac{1}{2} \frac{\gamma+1}{\gamma-1}}$$

\*Find  $\frac{A_e}{A^*}$  given Exit Mach number or solve numerically to find Exit Mach from  $\frac{A_e}{A^*}$   
 \*Once  $M_e$  is known, can solve or  $T_e$  and  $p_e$  using isentropic relations (above).

Nozzle Choking Criterion:

$$\frac{p_t}{p^*} = \left( \frac{\gamma+1}{2} \right)^{\frac{\gamma}{\gamma-1}} \text{ at } M^* = 1$$

\*Unchoked if,

$$\frac{p_t}{p^*} < \left( \frac{\gamma+1}{2} \right)^{\frac{\gamma}{\gamma-1}}$$

Exit Velocity:

$$V_e = a_e M_e \text{ where, } a_e = \sqrt{\gamma R T_e}$$

$$V_e = \left\{ \frac{2\gamma}{\gamma-1} R T_t \left[ 1 - \left( \frac{p_e}{p_t} \right)^{\frac{\gamma-1}{\gamma}} \right] \right\}^{\frac{1}{2}}$$

Where,

$$R = \frac{\bar{R}}{MW} = \left( \frac{8314 \text{ kJ}}{\text{mol} \cdot \text{K}} \right) \frac{1}{MW}$$

\*Therefore, lower MW gives higher  $V_e \rightarrow$  higher thrust

Isentropic Model of Atmosphere:

$$\frac{p(z)}{p_s} \approx \left[ 1 - \frac{\gamma-1}{2} \left( \frac{z}{z^*} \right)^2 \right]^{\frac{\gamma}{\gamma-1}}$$

\*Where surface pressure,  $p_s = 101.3 (10^3) \frac{N}{m^2}$   
 $z^* = 8404 \text{ m}$

Summerfeld Separation Criterion:

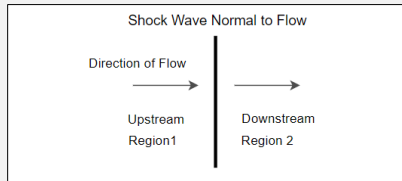
When nozzle flow is "highly" over expanded, when  $\frac{p_e}{p_\infty}$  gets too small, flow in nozzle separates when,

$$\frac{p_e}{p_\infty} \lesssim K$$

for  $0.25 \leq K \leq 0.4$

\*Boundary layer separation can cause highly turbulent recirculating flow

Normal Shock Relations: (M = Mach before shock)



Total Relations:

$$\frac{p_{t2}}{p_{t1}} = \left[ \frac{(\gamma+1) M^2}{(\gamma-1) M^2 + 2} \right]^{\frac{\gamma}{\gamma-1}} \left[ \frac{\gamma+1}{2\gamma M^2 - (\gamma-1)} \right]^{\frac{1}{\gamma-1}}$$

$$\frac{T_{t2}}{T_{t1}} = 1$$

Shock Jump Relations:

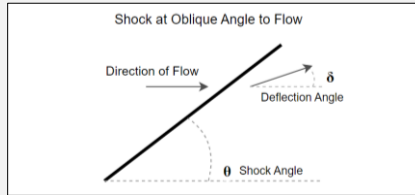
$$\frac{p_2}{p_1} = \frac{2\gamma M^2 - (\gamma-1)}{\gamma+1}$$

$$\frac{T_2}{T_1} = \frac{[2\gamma M^2 - (\gamma-1)][(\gamma-1)M^2 + 2]}{(\gamma+1)^2 M^2}$$

$$\frac{\rho_2}{\rho_1} = \frac{(\gamma+1)M^2}{(\gamma-1)M^2 + 2}$$

$$M_2^2 = \frac{(\gamma-1)M_1^2 + 2}{2\gamma M_1^2 - (\gamma-1)}$$

Oblique Shock Jump Relations:



Angle-Mach Relations:

$$\cot(\delta) = \tan(\theta) \left[ \frac{(\gamma+1)M_1^2}{2(M_1^2 \sin^2(\theta) - 1)} - 1 \right]$$

$$M_2^2 \sin^2(\theta - \delta) = \frac{(\gamma-1)M_1^2 \sin^2(\theta) + 2}{2\gamma M_1^2 \sin^2(\theta) - (\gamma-1)}$$

Total Relations:

$$\frac{p_{t2}}{p_{t1}} = \left[ \frac{(\gamma+1) M_1^2 \sin^2(\theta)}{(\gamma-1) M_1^2 \sin^2(\theta) + 2} \right]^{\frac{\gamma}{\gamma-1}} \left[ \frac{\gamma+1}{2\gamma M_1^2 \sin^2(\theta) - (\gamma-1)} \right]^{\frac{1}{\gamma-1}}$$

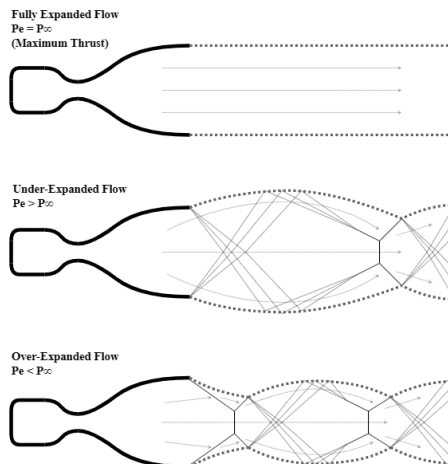
Shock Jump Relations:

$$\frac{p_2}{p_1} = \frac{2\gamma M_1^2 \sin^2(\theta) - (\gamma-1)}{\gamma+1}$$

$$\frac{T_2}{T_1} = \frac{[2\gamma M_1^2 \sin^2(\theta) - (\gamma-1)][(\gamma-1)M_1^2 \sin^2(\theta) + 2]}{(\gamma+1)^2 M_1^2 \sin^2(\theta)}$$

$$\frac{\rho_2}{\rho_1} = \frac{(\gamma+1)M_1^2 \sin^2(\theta)}{(\gamma-1)M_1^2 \sin^2(\theta) + 2}$$

Nozzle Expansion:



Thrust Performance

Thrust Equation:

$$T = \dot{m}_p V_e + (p_e - P_\infty) A_e$$

\*First term is jet (momentum) thrust, the second term is pressure thrust.

$$T = \dot{m}_p V_{eq}$$

\*Valid in all cases

Equivalent Velocity:

$$V_{eq} = V_e + \frac{(p_e - p_\infty) A_e}{\dot{m}_p} = \frac{T}{\dot{m}_p}$$

\*Would be the exit velocity from a fully expanded nozzle that produces same thrust as actual nozzle.

Specific Impulse:

$$I_{sp} = \frac{V_{eq}}{g_0} = \frac{T}{\dot{m}_p g_0}$$

\*Equivalent time that 1 lbf of combustion products could produce 1 lbf of thrust, units of seconds

Thrust Coefficient:

$$(c_T)_{actual} = \frac{T}{p_t A^*} \quad (c_T)_{isentropic} =$$

$$\gamma \left\{ \left( \frac{2}{\gamma-1} \right) \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}} \left[ 1 - \left( \frac{p_e}{p_t} \right)^{\frac{\gamma-1}{\gamma}} \right] \right\}^{\frac{1}{2}} + \frac{(p_e - p_\infty) A_e}{p_t A^*}$$

$$(c_T)_{ideal} = \gamma \left\{ \left( \frac{2}{\gamma-1} \right) \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}} \left[ 1 - \left( \frac{p_e}{p_t} \right)^{\frac{\gamma-1}{\gamma}} \right] \right\}^{\frac{1}{2}}$$

\*Isentropic and fully expanded nozzle flow ( $p_e = p_\infty$ )

$$(c_T)_{ideal vacuum} = \gamma \left\{ \left( \frac{2}{\gamma-1} \right) \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}} \right\}^{\frac{1}{2}}$$

\*Fully expanded flow in a vacuum (infinite nozzle length)

Trajectories

Tsiolkovsky Rocket Equation (velocity increment):

$$\Delta V_{burn} = V_{eq} \ln \left( \frac{M_0}{M_f} \right) - g_0 t_b$$

- Larger velocity increment due to high thrust for short burn time than due to lower thrust for long burn time.  
 - Short burn at high thrust reduces energy consumed lifting propellant (for vertical launch)

Aerodynamic Drag:

$$D = \frac{1}{2} \rho V^2 A C_D$$

\*Circular cross-section with diameter d has area  $A = \frac{\pi}{4} d^2$

Gravitational Force:

$$g(z) = g_e \left( \frac{R_e}{R_e + z} \right)^2$$

\*As a function of altitude z in m, earth radius  $R_e = 6378 (10^3) \text{ m}$ , and acceleration of gravity at surface  $g_e = 9.8 \text{ m/s}^2$

Time and Altitude at Burnout:

$$h_b = V_{eq} \left\{ 1 - \frac{\ln(R)}{R-1} \right\} t_b - \frac{1}{2} g_e t_b^2$$

\*First term is dependent on Mass Ratio  $R = \left( \frac{M_0}{M_f} \right)$

$$t_b = \left( \frac{M_p}{\dot{m}_p} \right) \quad t_b = (t - t_0)$$

Maximum Vehicle Altitude (vertical launch):

$$h_{max} = h_b + \frac{1}{2} \frac{(\Delta V_b)^2}{g_e}$$

$$h_{max} = \frac{V_{eq}^2 (\ln(R))^2}{2g_e} - V_{eq} t_b \left\{ \frac{R}{R-1} \ln(R) - 1 \right\}$$

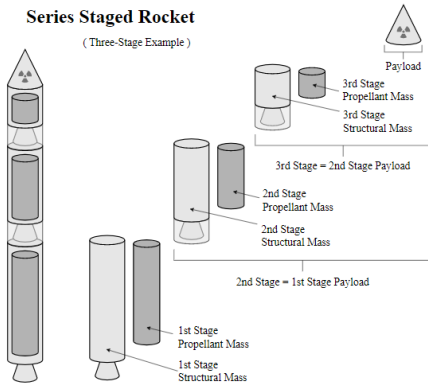
\*Minimizing  $t_b$  maximizes  $h_{max}$

## Rocket Staging

### Staging Methods:

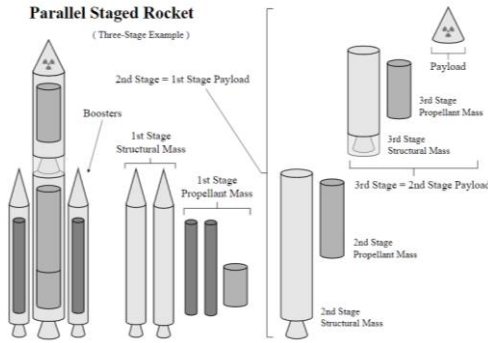
#### Series Staged Rocket

(Three-Stage Example)



#### Parallel Staged Rocket

(Three-Stage Example)



1 <sup>st</sup> Stage = $M_0$	2 <sup>nd</sup> Stage = $M_{02}$	3 <sup>rd</sup> Stage = $M_{03}$
		$M_L = \text{Payload (final)}$
$M_{L1} = \text{Payload} =$	$M_{L2} = \text{Payload} =$	$M_{S3} = \text{Structural}$
	$M_{S2} = \text{Structural}$	$M_{P3} = \text{Propellant}$
	$M_{P2} = \text{Propellant}$	
$M_{S1} = \text{Structural}$		
$M_{P1} = \text{Propellant}$		

\*For Parallel Staging: Mass of 1<sup>st</sup> stage propellant from large tank is equal to the total mass of the large tank's propellant times the fraction of the 1<sup>st</sup> stage burn time over the total burn time of the 1<sup>st</sup> and 2<sup>nd</sup> stages.

**Stage Mass Relation:**  $N$ -stage vehicle, the  $i^{\text{th}}$ -stage has,  
 $M_{0i} = M_{Pi} + M_{Si} + M_{Li}$

#### Payload Ratio:

$$\lambda_i \equiv \frac{M_{0(i+1)}}{M_{0i} - M_{0(i+1)}} = \frac{M_{Li}}{M_{0i} - M_{Li}} = \frac{M_{Li}}{M_{Pi} + M_{Si}}$$

\*Equal for all similar stages,  $\lambda_i = \lambda_{(i+1)} = \dots = \lambda_N$

$$\lambda_{\text{optimal}} = \frac{(M_L/M_0)^{1/N}}{(1 - M_L/M_0)^{1/N}}$$

#### Structural Coefficient:

$$\varepsilon_i = \frac{M_{Si}}{M_{0i} - M_{0(i+1)}} = \frac{M_{Si}}{M_{0i} - M_{Li}} = \frac{M_{Si}}{M_{Si} + M_{Pi}}$$

\*Equal for all similar stages,  $\varepsilon_i = \varepsilon_{(i+1)} = \dots = \varepsilon_N$

#### Mass Ratio:

$$R_i = \frac{M_{0i}}{M_{0i} - M_{Pi}} = \frac{M_{0i}}{M_{Si} + M_{Li}} = \frac{1 + \lambda_i}{\varepsilon_i + \lambda_i}$$

### Velocity Increments for an Optimally Staged Rocket:

$\Delta V$  due to burnout of the  $i^{\text{th}}$ -stage,

$$(V_b)_i = (V_{eq})_i \ln(R_i) - g_0(t_b)_i$$

Final  $\Delta V$  imparted on final payload  $M_L$  of stage  $N$ ,

$$T_b = \sum_{i=1}^N (t_b)_i$$

$$(V_b)_N = \sum_{i=1}^N \left[ (V_{eq})_i \ln \left( \frac{1 + \lambda_i}{\varepsilon_i + \lambda_i} \right) \right] - g_0 T_b$$

$$(\Delta V_b)_N = V_{eq} N \ln \left\{ \varepsilon + (1 - \varepsilon) \left[ \frac{M_L}{M_0} \right]^{\frac{1}{N}} \right\}^{-1} - g_0 T_b$$

Ideal Case (infinitely many stages):

$$(\Delta V_b)_{N \rightarrow \infty} = V_{eq} (1 - \varepsilon) \ln \left[ \frac{M_0}{M_L} \right] - g_0 T_b$$

\*Equivalent Velocity  $V_{eq} = \frac{T}{m_p} \approx$  same for all stages

### Orbital Dynamics

#### Gravitational Parameter:

$$\mu = GM' \quad \text{where } G = 6.67 \times 10^{-11} \frac{N \cdot m^2}{kg}$$

#### Radial Position (from vehicle to center of planet):

$$r = \frac{a(1 - \varepsilon^2)}{1 + \varepsilon \cos(\theta)}$$

#### Semimajor Axis:

$$a = r_{\min} \frac{1}{1 - \varepsilon} \quad a = \frac{1}{2} (r_a + r_p)$$

#### Eccentricity:

$$\varepsilon = \frac{r_a - r_p}{r_a + r_p} \quad \varepsilon = \sqrt{1 - \frac{b^2}{a^2}} \quad \varepsilon = 1 - \frac{r_{\min}}{a}$$

\*Where  $b$  = semi-minor axis

$\varepsilon < 1 \rightarrow$  ellipse

$\varepsilon = 1 \rightarrow$  parabola

$\varepsilon > 1 \rightarrow$  hyperbola

$\varepsilon = 0 \rightarrow$  circle

#### Vis-Viva:

$$V^2 = \mu \left( \frac{2}{r} - \frac{1}{a} \right)$$

#### Perigee Radius:

$$r_p = a(1 - \varepsilon)$$

#### Perigee Velocity:

$$V_p = \sqrt{\frac{\mu}{a} \left( \frac{1 + \varepsilon}{1 - \varepsilon} \right)} \quad V_p = \sqrt{\frac{\mu}{r_p} \frac{2r_a}{r_a + r_p}}$$

#### Apogee Radius:

$$r_a = a(1 + \varepsilon)$$

#### Apogee Velocity:

$$V_a = \sqrt{\frac{\mu}{a} \left( \frac{1 - \varepsilon}{1 + \varepsilon} \right)} \quad V_a = \sqrt{\frac{\mu}{r_a} \frac{2r_p}{r_a + r_p}} \quad V_a = \sqrt{\frac{\mu r_a}{a r_p}}$$

#### Orbital Period:

$$T = 2\pi \sqrt{\frac{a^3}{\mu}}$$

#### Energy (per unit mass):

$$e = -\frac{\mu}{2a}$$

### Circular Orbits: $r(\theta) = R = \text{constant}$ , $e < 0$ , $\varepsilon = 0$

Velocity:

$$V = \sqrt{\frac{\mu}{R}}$$

Period:

$$T = 2\pi \sqrt{\frac{R^3}{\mu}}$$

Escape from Circular Orbit:

$$\Delta V_1 \geq (\sqrt{2} - 1) \sqrt{\frac{\mu}{R}}$$

### Parabolic Trajectories: (not orbit) $e = 0$ , $a = \infty$ , $\varepsilon = 1$

$$V_{\infty} = 0$$

Escape Velocity:

$$V_{\text{esc}} = \sqrt{\frac{2\mu}{r}}$$

Escape from planet surface:

$$V_{\text{esc}} = \sqrt{\frac{2\mu}{R_{\text{planet}}}}$$

### Hyperbolic Trajectories: (not orbit) $e > 0$

Excess Velocity:

$$V_{\infty} = \sqrt{2e} \quad V_{\infty} = \sqrt{\frac{\mu}{-a}}$$

Asymptote Angle:

$$\theta_{\infty} = \pi - \cos^{-1} \left( \frac{1}{\varepsilon} \right)$$

Turning Angle:

$$\delta = 2 \sin^{-1} \left( \frac{1}{\varepsilon} \right)$$

Miss Distance:

$$\Delta = -a\sqrt{\varepsilon^2 - 1}$$

### Orbital Maneuvers

#### Circularization Burn:

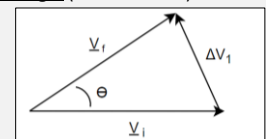
Circularize at  $r_a$ :

$$\Delta V_1 = \sqrt{\frac{\mu}{r_a}} \left[ 1 - \sqrt{\frac{2r_p}{r_a + r_p}} \right]$$

Circularize at  $r_p$ :

$$\Delta V_1 = \sqrt{\frac{\mu}{r_p}} \left[ 1 - \sqrt{\frac{2r_a}{r_a + r_p}} \right]$$

#### Inclination Change: (circular orbits)



\*Both other angles are:  $\frac{1}{2}\pi - \theta$

$$\Delta V_1 = 2 \sqrt{\frac{\mu}{R}} \sin \left( \frac{\theta}{2} \right)$$

#### Orbital Rendezvous:

Lead Angle:

$$\alpha_L = \frac{\pi}{2} \sqrt{\frac{(R_1 + R_2)^3}{2(R_2)^3}}$$

Transfer Time:

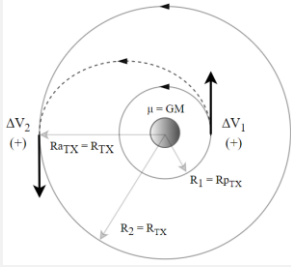
$$T_{TX} = T_{\text{Hohmann}} = \frac{\pi}{2} \sqrt{\frac{(R_1 + R_2)^3}{2\mu}}$$

Intercept Opportunity Times:

$$\frac{1}{T} = \frac{1}{(T_C)_1} - \frac{1}{(T_C)_2} \quad \text{where } (T_C)_i = 2\pi \sqrt{\frac{R_i}{\mu}}$$

## Orbital Transfers

### Hohmann Transfer:



Circular Orbit Velocities:

$$(V_c)_1 = \sqrt{\frac{\mu}{R_1}} \quad (V_c)_2 = \sqrt{\frac{\mu}{R_2}}$$

Transfer Orbit Parameters: (Trans time, see Rendezvous)

$$r_p = R_1$$

$$r_a = R_2$$

$$V_p = \sqrt{\frac{\mu}{R_1} \frac{2R_2}{R_1 + R_2}}$$

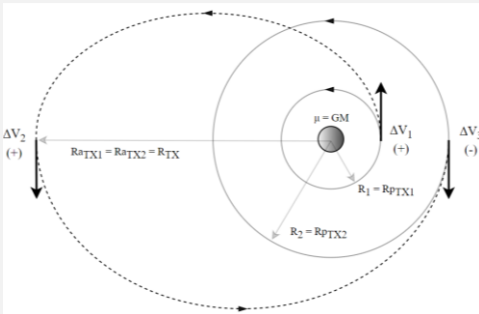
$$V_a = \sqrt{\frac{\mu}{R_2} \frac{2R_1}{R_1 + R_2}}$$

Velocity Increments:

$$\Delta V_1 = \sqrt{\frac{\mu}{R_1} \left[ \sqrt{\frac{2R_2}{R_1 + R_2}} - 1 \right]}$$

$$\Delta V_2 = \sqrt{\frac{\mu}{R_2} \left[ 1 - \sqrt{\frac{2R_1}{R_1 + R_2}} \right]}$$

### Bielliptic Transfer:



Circular Orbit Velocities:

$$(V_c)_1 = \sqrt{\frac{\mu}{R_1}} \quad (V_c)_2 = \sqrt{\frac{\mu}{R_2}}$$

Transfer Orbit Parameters: ( $R_{TX}$  can be chosen)

$$R_1 = R_{pTX1}$$

$$R_{TX} = R_{aTX1}$$

$$R_2 = R_{pTX2}$$

$$R_{TX} = R_{aTX2}$$

$$a_{TX1} = \frac{1}{2}(R_1 + R_{TX})$$

$$a_{TX2} = \frac{1}{2}(R_2 + R_{TX})$$

Transit Time:

$$T_{TX} = \pi \frac{\sqrt{(a_{TX1})^3} + \sqrt{(a_{TX2})^3}}{\sqrt{\mu}}$$

Velocity Increments:

$$\Delta V_1 = \sqrt{\mu \left( \frac{2}{R_1} - \frac{1}{a_{TX1}} \right)} - \sqrt{\frac{\mu}{R_1}}$$

$$\Delta V_2 = \sqrt{\mu \left( \frac{2}{R_{TX}} - \frac{1}{a_{TX2}} \right)} - \sqrt{\mu \left( \frac{2}{R_{TX}} - \frac{1}{a_{TX1}} \right)}$$

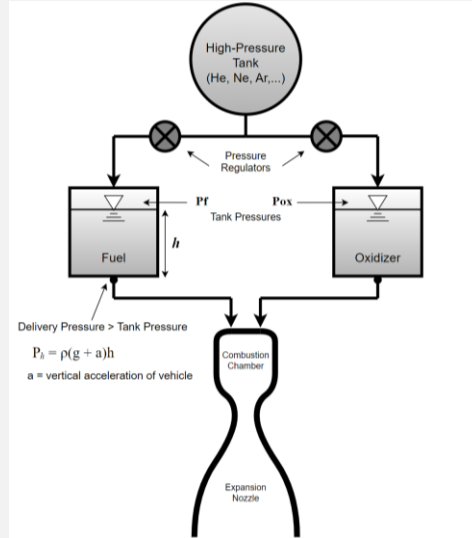
$$\Delta V_3 = \sqrt{\frac{\mu}{R_2}} - \sqrt{\mu \left( \frac{2}{R_2} - \frac{1}{a_{TX2}} \right)}$$

### Standard Gravitation Parameters

Body	$\mu = GM \left[ \frac{m^3}{s^2} \right]$
Sun	$1.327 \times 10^{20}$
Mercury	$2.203 \times 10^{13}$
Venus	$3.249 \times 10^{14}$
Earth	$3.986 \times 10^{14}$
(Moon)	$4.903 \times 10^{12}$
Mars	$4.283 \times 10^{13}$
Jupiter	$1.267 \times 10^{17}$
Saturn	$3.793 \times 10^{16}$
Uranus	$5.794 \times 10^{15}$
Neptune	$6.837 \times 10^{15}$
Pluto	$1.108 \times 10^{12}$

### Propellant Feed Systems

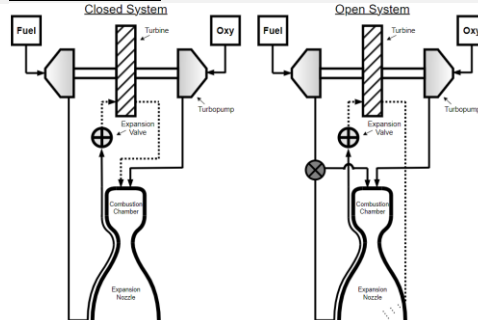
#### Pressure-Fed Systems:



Deliver Pressure > Tank Pressure,  
 $P_h = \rho(g + a)h$

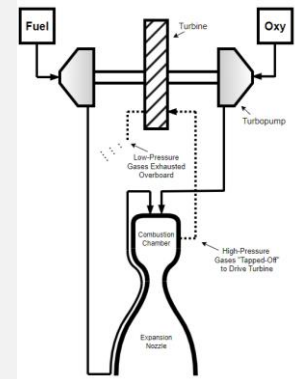
	$\rho$ [kg/m <sup>3</sup> ]	$C_V$ [J/kg · K]	$T$ [K]
LO2	1141	1669	90
LH2	70.8	9668	20
LCH4	422	4216	111
RP-1	810	2188	300

#### Expander Systems:



\*Fuel is expanded to gaseous form to drive turbine that powers pumps which then push propellant into CC.  
 - Low/moderate thrust applications  
 - Often used for upper stage engines

### Combustion Tap-Off Systems:



\*Uses combustion product gas from combustion chamber to drive gas turbine that then drives pumps.

#### Open Systems:

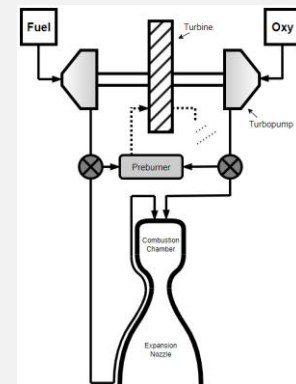
- Turbine exit gas pressure low, dumped overboard

#### Closed Systems:

- Turbine exit gas dumped into nozzle at low pressure location to add  $\dot{m}_p$  at nozzle exit

\*Moderate/high thrust (Saturn I-B/Blue Origin)

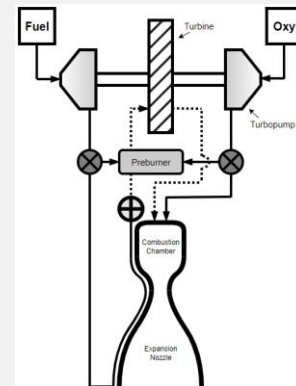
### Gas Generator Systems:



\*Part of fuel and oxidizer are burned in preburner to produce high-pressure gas that drives turbopumps.

- Preburner operated very fuel-rich to keep temperature sufficiently low
- Fuel-rich exhaust dumped overboard (open) or dumped into nozzle (closed)
- Power controlled by adjusting fuel/ox rates into preburner
- Can achieve higher CC pressure than expander sys.

### Staged Combustion Systems:



\*All propellant is burned in two stages (nothing wasted). First in fuel-rich preburner, then in main CC.

- Allows very high CC pressures (high thrust)
- Can be used with any liquid propellants
- More efficient than gas generator (all fuel burned)
- Requires very high-pressure turbopumps

## Propellants and Aerothermochemistry

### Propellant Types:

#### Liquid Propellants:

##### Fuels:

##### a) Cryogenic

- LH<sub>2</sub> – liquid hydrogen (20 K) “deep cryogen”
- LCH<sub>4</sub> – liquid methane (111 K)

##### b) Non-cryogenic

- RP-1 – kerosene [C<sub>1</sub>H<sub>1.96</sub>]
- ethyl alcohol – [C<sub>2</sub>H<sub>5</sub>OH]

##### c) Hypergolic (E<sub>a</sub> ≤ 0) (reacts on contact with oxi.)

- hydrazine [N<sub>2</sub>H<sub>4</sub>]
- monomethyl hydrazine (MMH)
- unsymmetrical dimethyl hydrazine (UDMH)

##### Oxidizers:

##### a) Cryogenic

- LOX – liquid oxygen
- LF<sub>2</sub> – liquid fluorine
- FLOX – fluorine-enhanced LOX
- N<sub>2</sub>O<sub>4</sub> – dinitrogen tetroxide

##### b) Non-cryogenic

- N<sub>2</sub>O – nitrous oxide
- N<sub>2</sub>O<sub>4</sub> – dinitrogen tetroxide
- ClF<sub>5</sub> – chlorine pentafluoride

#### Monopropellants:

\* Both fuel and oxidizer are contained in the same molecule; contact with a catalyst bed initiates reaction (breaks oxidizer and fuel components apart).

\* Usually used for RCS and in-space propulsion at low-mid thrust applications.

- hydrazine (N<sub>2</sub>H<sub>4</sub>) + granular Al w/iridium coating
- hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) + many possible catalysts
- nitromethane (CH<sub>3</sub>NO<sub>2</sub>)
- ethylene oxide (C<sub>2</sub>H<sub>4</sub>O)
- nitrous oxide (N<sub>2</sub>O)
- hydroxylammonium nitrate (HAN) (H<sub>2</sub>N<sub>2</sub>O<sub>4</sub>)

#### Solid Propellants:

\* Both fuel and oxidizer are initially in solid form.

##### Homogenous Propellants:

\* Fuel and oxidizer are contained in the same molecule or in a homogeneous mixture.

- nitroglycerine + nitrocellulose  
C<sub>3</sub>H<sub>5</sub>(NO<sub>2</sub>)<sub>3</sub> + C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>(NO<sub>2</sub>)<sub>3</sub>
- Single/Double/Triple base propellants
- Metal powders often added to increase heat of combustion.

##### Heterogeneous (composite) Propellants:

\* Heterogeneous mixture of oxidizing crystals held in an organic plastic-like fuel “binder”

##### a) Fuel Component (common binders)

- hydroxyl-terminated polybutadiene (HTPB)
- rubber/asphalt

##### b) Oxidizer Component (ground crystals of one or more of the following)

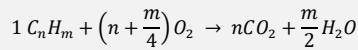
- ammonium perchlorate (AP)
- ammonium nitrate (AN)
- nitronium perchlorate (NP)
- potassium perchlorate (KP)
- potassium nitrate (PN)
- cyclotrimethylene trinitomine (RDX)
- cyclotetramethylene tetranitramine (HMX)

\* The combination of fuel/ox specifies the propellant (e.g. AP-HTPB, KP-HTPB, ...)

Many possible combinations.

### Hydrocarbon Combustion:

Overall Stoichiometric Reaction:



$n = \text{carbon atoms}, m = \text{hydrogen atoms}$

Mixture Ratio:

$$r = \left(\frac{M_{O_2}}{M_f}\right)$$

Stoichiometric Mixture Ratio:

$$r_s = \frac{32n + 8m}{12n + m}$$

Equivalence Ratio:

$$\phi = \frac{r_s}{r}$$

- $\phi = 1 \rightarrow \text{stoichiometric combustion}$
- $\phi > 1 \rightarrow \text{fuel-rich combustion}$
- $\phi < 1 \rightarrow \text{fuel-lean combustion}$

Molar Enthalpy of Combustion:

$$\Delta \tilde{h}_c = \left[ \sum_i \text{Bond Energy} \right]_{\text{react}} - \left[ \sum_i \text{Bond Energy} \right]_{\text{prod}}$$

$\Delta \tilde{h}_c < 0 \rightarrow \text{exothermic}$

$\Delta \tilde{h}_c > 0 \rightarrow \text{endothermic}$

Convert to mass-specific enthalpy:

$$\Delta h_c = \frac{\Delta \tilde{h}_c}{MW_{HC}}$$

$$MW_{HC} = 12n + m \left[ \frac{\text{kcal}}{g} \right]$$

Convert to [kJ/kg]:

$$\Delta h_c \left[ \frac{\text{kcal}}{g} \right] \cdot \left( \frac{1000 \text{ cal}}{\text{kcal}} \right) \left( \frac{1000 g}{kg} \right) \left( \frac{4.1814 J}{\text{cal}} \right) \left( \frac{1 kJ}{1000 J} \right)$$

### Turbopumps and Turbines

#### Turbopumps:

Shaft Power from Turbine:

$$\dot{W} = \tau \cdot \dot{\Omega}$$

Ideal Pump Work:

$$|w|_{\text{ideal}} = \frac{1}{\rho} (p_{t2} - p_{t1})$$

Ideal Pump Power:

$$|\dot{w}|_{\text{ideal}} = \dot{m} |w|_{\text{ideal}} = \dot{m} \frac{1}{\rho} (p_{t2} - p_{t1})$$

Pump Efficiency:

$$\eta_p = \frac{|w|_{\text{ideal}}}{|w|_{\text{actual}}} = \frac{|\dot{w}|_{\text{ideal}}}{|\dot{w}|_{\text{actual}}}$$

Non-Ideal Pump Work:

$$|w|_{\text{actual}} = \frac{1}{\eta_p} \frac{1}{\rho} (p_{t2} - p_{t1})$$

Non-Ideal Pump Power:

$$|\dot{w}|_{\text{actual}} = \dot{m} |w|_{\text{actual}} = \dot{m} \frac{1}{\eta_p} \frac{1}{\rho} (p_{t2} - p_{t1})$$

\* from the above

$$\Delta p_t = (p_{t2} - p_{t1}) = w_p \eta_p \frac{\rho}{\dot{m}}$$

1<sup>st</sup> Law for Liquid Flow Through Pump:

$$c_v = (T_2 - T_1) + \frac{1}{\rho} (p_2 - p_1) + \frac{1}{2} (V_2^2 - V_1^2) = |w|$$

$$c_v = (T_2 - T_1) + \frac{1}{\rho} (p_{t2} - p_{t1}) = |w|$$

Temperature Change Across Pump:

$$(T_2 - T_1) = \left( \frac{1 - \eta_p}{\eta_p} \right) \frac{(p_{t2} - p_{t1})}{\rho \cdot c_v}$$

### Turbines:

Application of 1<sup>st</sup> Law:

$$(\Delta h_t)_{1,2} = q_{1,2} - w_{1,2} = 0$$

$$\rightarrow h_{t2} - h_{t1} = 0$$

$$\rightarrow C_p (T_{t2} - T_{t1}) = 0$$

$$\therefore T_{t2} = T_{t1}$$

Turbine Work:

$$|W|_{\text{nonideal}} = \eta_T \frac{1}{\rho} \Delta p_t$$

Temperature Change Across Turbine:

$$\Delta T = (1 - \eta_T) \frac{\Delta p_t}{\rho \cdot c_v}$$

Turbine Work: (per unit mass)

$$|w_T| = |C_p (T_{te} - T_{ti})|$$

Turbine Power:

$$\dot{W}_T = \dot{m} |w_T|$$

Turbine Stage Efficiency:

$$\eta_s = \frac{|w|}{|w|_s}$$

Total Turbine Efficiency:

$$\eta_T = \frac{1 - \left(\frac{T_{t3}}{T_{t1}}\right)}{1 - \left(\frac{p_{t3}}{p_{t1}}\right)^{\frac{\gamma-1}{\gamma}}}$$

Turbine Total Temperature Ratio:

$$\frac{T_{t3}}{T_{t1}} = 1 - \eta_T \left[ 1 - \left(\frac{p_{t3}}{p_{t1}}\right)^{\frac{\gamma-1}{\gamma}} \right]$$

Turbine Total Pressure Ratio:

$$\frac{p_{t3}}{p_{t1}} = \left[ 1 - \frac{1}{\eta_T} \left( 1 - \frac{T_{t3}}{T_{t1}} \right) \right]^{\frac{\gamma}{\gamma-1}}$$

Total-to-Static Temperature Relation:

$$T_3 = T_{t3} \left[ 1 + \frac{\gamma-1}{2} M_3^2 \right]^{-1}$$

Total-to-Static Pressure Relation:

$$p_3 = p_{t3} \left[ 1 + \frac{\gamma-1}{2} M_3^2 \right]^{\frac{\gamma}{\gamma-1}}$$

Entropy Change Across Turbine:

$$\Delta s = s_3 - s_1 = C_p \ln \left( \frac{T_{t3}}{T_{t1}} \right) - R \ln \left( \frac{p_{t3}}{p_{t1}} \right)$$

$$R = C_p \left( \frac{\gamma-1}{\gamma} \right)$$

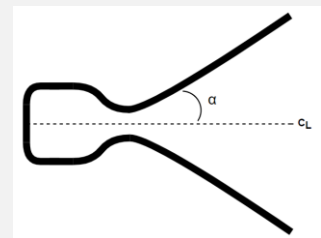
$$R = C_p (\gamma - 1)$$

Temperature-Velocity Relation:

$$(T_t - T) = \frac{V^2}{2 \cdot C_p}$$

### Nozzle Design

#### Simple Conical Nozzles:



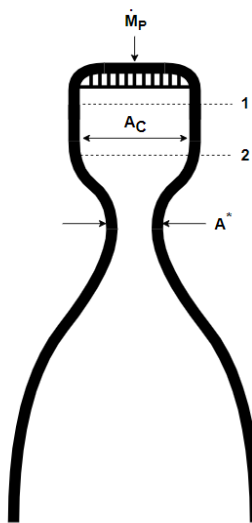
Axial Thrust:

$$T_{axial} = \frac{1 + \cos(\alpha)}{2} \dot{m}_p V_e + (p_e - p_\infty) A_e$$

\*See notes for Rao and TOP nozzle stuff

## Combustion Chamber Performance

### Combustion Chambers:



Thrust Coefficient:

$$C_T = \frac{T}{p_{t2} \cdot A^*}$$

Thrust:

$$T = \dot{m}_p \left( \frac{p_{t2} \cdot A^*}{\dot{m}_p} \right) C_T$$

C\* "Characteristic Velocity":

$$C^* = \frac{p_{t2} \cdot A^*}{\dot{m}_p}$$

$$C_{ideal}^* = \frac{p_{t2}}{p_{t1}} \sqrt{RT_{t2}} \left[ \gamma \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{2}} \right]^{\frac{1}{2}}$$

C\* Efficiency:

$$\eta_{c^*} = \frac{C_{actual}^*}{C_{ideal}^*}$$

Specific Impulse:

$$I_{sp} = C^* C_T \frac{1}{g}$$

Pressure Ratio:

$$\frac{p_2}{p_1} = \frac{(1 + \gamma M_1^2)}{(1 + \gamma M_2^2)}$$

Temperature Ratio:

$$\frac{T_2}{T_1} = \frac{[1 + \gamma M_1^2]}{[1 + \gamma M_2^2]} \cdot \left( \frac{M_2}{M_1} \right)^2$$

Total Temperature Ratio:

$$\frac{T_{t2}}{T_{t1}} = \frac{[1 + \frac{\gamma-1}{2} M_2^2]}{[1 + \frac{\gamma-1}{2} M_1^2]} \cdot \left( \frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right) \left( \frac{M_2}{M_1} \right)^2$$

$$\frac{T_{t2}}{T_{t1}} = \frac{q_{1,2}}{C_p T_{t1}} + 1$$

Total Pressure Ratio:

$$\frac{p_{t2}}{p_{t1}} = \frac{[1 + \frac{\gamma-1}{2} M_2^2]^{\frac{\gamma}{\gamma-1}}}{[1 + \frac{\gamma-1}{2} M_1^2]^{\frac{\gamma}{\gamma-1}}} \cdot \left( \frac{1 + \gamma M_1^2}{1 + \gamma M_2^2} \right)$$

Combustion Chamber-to-Throat Area Ratio:

$$\frac{A_c}{A^*} = \frac{1}{M_2} \left\{ \left( \frac{2}{\gamma+1} \right) \left[ 1 + \frac{\gamma-1}{2} M_2^2 \right] \right\}^{\frac{\gamma+1}{2\gamma-1}}$$

Thermal Choking Criterion:

$$\frac{T_{t2}}{T_{t1}} \geq \frac{1}{2(1+\gamma)} \frac{(1 + \gamma M_1^2)^2}{[1 + \frac{\gamma-1}{2} M_1^2] M_1^2}$$

\*subtract 1 from result to get max allowable  $T_{t2}$

## DOF and Molecular Energy

Monatomic Molecules:

$$\begin{aligned} (DOF)_{trans} &= 3 \\ (DOF)_{vib} &= 0 \\ (DOF)_{rot} &= 0 \\ \rightarrow (DOF)_{avail} &= 3 \end{aligned}$$

Diatomic Molecules:

$$\begin{aligned} (DOF)_{trans} &= 3 \\ (DOF)_{vib} &= 2 \\ (DOF)_{rot} &= 2 \\ \rightarrow (DOF)_{avail} &= 7 \end{aligned}$$

Polyatomic Molecules:

$$\begin{aligned} (DOF)_{trans} &= 3 \\ (DOF)_{vib} &= \begin{cases} 2 & \text{if linear (CO}_2\text{)} \\ 3 & \text{if nonlinear (all others)} \end{cases} \\ (DOF)_{rot} &= \begin{cases} 2(3n-5) & \text{if linear} \\ 2(3n-6) & \text{if nonlinear} \end{cases} \\ \rightarrow (DOF)_{avail} &= \begin{cases} 5 + 2(3n-5) & \text{if linear} \\ 6 + 2(3n-6) & \text{if nonlinear} \end{cases} \end{aligned}$$

Energy Stored in One Molecule:

$$e' = \frac{1}{2} kT (DOF)_{active}$$

$$k = \frac{\bar{R}}{A_v} = \frac{8.314 \text{ J/mol} \cdot \text{K}}{6.022 \times 10^{23} \frac{\text{part}}{\text{mol}}} \quad (\text{Boltzmann Constant})$$

$$\bar{e} = A_v \cdot e' = \frac{1}{2} \bar{R} T \cdot (DOF)_{active} \quad (\text{per mole})$$

$$e = \bar{e} / MW = \frac{1}{2} \bar{R} T \cdot (DOF)_{active} \quad (\text{per mass})$$

\*(DOF)<sub>active</sub> depends on temperature

- Low T = only trans. active
- Med T = trans. and rot. active
- High T = trans., rot., and vib. are active

### Specific Heat and Gas Constant Relations:

Specific Heat - Constant Volume:

$$\bar{C}_v = \bar{R} \frac{1}{2} (DOF)_{active}$$

$$C_v = R \frac{1}{2} (DOF)_{active}$$

Specific Heat - Constant Pressure:

$$\bar{C}_p = \bar{R} \left[ 1 + \frac{1}{2} (DOF)_{active} \right]$$

$$C_p = R \left[ 1 + \frac{1}{2} (DOF)_{active} \right]$$

Gas Constant Relations:

$$R = C_p - C_v$$

$$\bar{R} = \bar{C}_p - \bar{C}_v$$

\*given  $C_p$  and  $\gamma$

$$R = C_p \left( 1 - \frac{C_v}{C_p} \right) = C_p \left( 1 - \frac{1}{\gamma} \right) = C_p \left( \frac{\gamma-1}{\gamma} \right)$$

\*given  $C_v$  and  $\gamma$

$$R = C_p - C_v = C_v \left( \frac{C_p}{C_v} - 1 \right) = C_v (\gamma - 1)$$

Specific Heat Ratio:

$$\gamma = 1 + \frac{2}{(DOF)_{active}}$$

Monatomic:

$$(DOF)_{active} = 3 \quad (\text{independent of temp.})$$

$$\rightarrow \gamma = 1 + \frac{2}{3} = 1.667$$

Diatomic:

- Medium temp

$$(DOF)_{active} = 5 \quad \rightarrow \gamma = 1 + \frac{2}{5} = 1.4$$

- High temp

$$(DOF)_{active} = 7 \quad \rightarrow \gamma = 1 + \frac{2}{7} = 1.286$$

## Non-Isentropic Nozzle Flow

### Application of 1<sup>st</sup> Law:

$$\begin{aligned} (\Delta h_t)_{2,e} &= q_{2,e} - w_{2,e} = 0 \\ \rightarrow h_{t2} - h_{te} &= 0 \\ \rightarrow C_p (T_{t2} - T_{te}) &= 0 \\ \therefore T_{t2} &= T_{te} \end{aligned}$$

### Temperature-Velocity Relation:

$$(T_t - T) = \frac{V^2}{2 \cdot C_p}$$

### Total Pressure Ratio:

$$\frac{p_{te}}{p_{t2}} = e^{-\frac{\Delta s}{R}}$$

\*where,  $p_{te} < p_{t2}$  and  $\Delta s > 0$

### Total-to-Static Pressure Ratio:

$$\frac{p_e}{p_{t2}} = \left\{ 1 - \frac{1}{\eta_N} \left[ \frac{(\gamma-1) M_e^2}{2 + (\gamma-1) M_e^2} \right] \right\}^{\frac{\gamma}{\gamma-1}}$$

### Nozzle Efficiency:

$$\eta_N = \frac{1 - \left[ 1 + \frac{\gamma-1}{2} M_e^2 \right]^{-1}}{1 - \left( \frac{p_e}{p_{t2}} \right)^{\frac{\gamma-1}{\gamma}}}$$

### Area-Mach Relation:

$$\frac{A_e}{A^*} =$$

$$\frac{1}{M_e} \left\{ \left( \frac{2}{\gamma+1} \right) \left[ 1 + \frac{\gamma-1}{2} M_e^2 \right] \right\}^{\frac{\gamma+1}{2\gamma-1}} \cdot \left\{ 1 + \left( 1 - \frac{1}{\eta_N} \right) \frac{\gamma-1}{2} M_e^2 \right\}^{-\frac{\gamma}{\gamma-1}}$$

\*from resulting  $M_e$ , get:

$\rightarrow p_e$  from total-to-static (above)

$$\rightarrow T_e = T_{te} \left[ 1 + \frac{\gamma-1}{2} M_e^2 \right]^{-1}$$

$$\rightarrow a_e = \sqrt{\gamma R T_e} \quad \text{where, } R = C_p \left( \frac{\gamma-1}{\gamma} \right)$$

$$\rightarrow V_e = M_e \cdot a_e$$

$$\rightarrow T = m_p V_e + (p_e - p_\infty) A_e$$

### Non-Isentropic Thrust Coefficient:

$$C_T = \gamma \left\{ \eta_n \left( \frac{2}{\gamma-1} \right) \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}} \left[ 1 - \left( \frac{p_e}{p_{t2}} \right)^{\frac{\gamma-1}{\gamma}} \right] \right\}^{\frac{1}{2}} + \frac{(p_e - p_\infty) A_e}{p_{t2} A^*}$$

### Injectors and Droplet Vaporization

#### Droplet Characterization:

Droplet Diameter Probability:

$$P(D) = \frac{\text{Prob} \left( D + \frac{1}{2} dD \right)}{dD}$$

\*Sauter Mean Diameter (SMD): the droplet diameter with same surface-to-volume ratio as the entire spray.

\*Weber Number: ratio of inertia (mass) to surface tension

\*Ohnesorge Number: ratio of friction to surface tension

$$\frac{SMD}{\ell} = c \cdot We^p \cdot Oh^q$$

$$(SMD)_{CC} = (SMD)_{lab} \cdot \left( \frac{\ell_{CC}}{\ell_{lab}} \right) \left( \frac{We_{CC}}{We_{lab}} \right)^p \left( \frac{Oh_{CC}}{Oh_{lab}} \right)$$

#### Droplet Vaporization:

Heat of Vaporization: (heat flux into droplet)

$$\dot{m}_v \cdot \Delta h_v = \dot{Q}_{in} \quad \text{where,}$$

$$\dot{Q}_{in} = -k \frac{dT}{dr} \Big|_{R(t)} \cdot 4\pi R^2$$

$$\dot{m}_v = -4\pi \rho_L R^2(t) \frac{dR}{dt}$$

Droplet Diameter: (as function of time)

$$D^2(t) = D_0 - \left( \frac{8k}{\rho_L C_p} \right) \ln(1+B) \cdot t$$

$$B \equiv \frac{C_p (T_\infty - T_v)}{\Delta h_v}$$

Vaporization time:

$$t_v = D_0^2 \frac{\rho_L C_p}{8k \ln(1+B)} = \frac{D_0^2}{k_v}$$